

Preparation of Optically Active (4 α ,8 α)-Octahydro-4 α -methyl-8-methylene-2(1H)-naphthalenone, a Key Intermediate for the Enantioselective Synthesis of Eudesmane Sesquiterpenes

Masanori UTAKA, Yasuyuki FUJII, and Akira TAKEDA*

Department of Synthetic Chemistry, School of Engineering,
Okayama University, Tsushima, Okayama 700

The titled optically active methylene ketone was prepared in 45% ee from 3-methyl-1,2-cyclohexanedione by use of a novel asymmetric Michael addition with (R,R)-(-)-2,3-butanediol as chiral auxiliary.

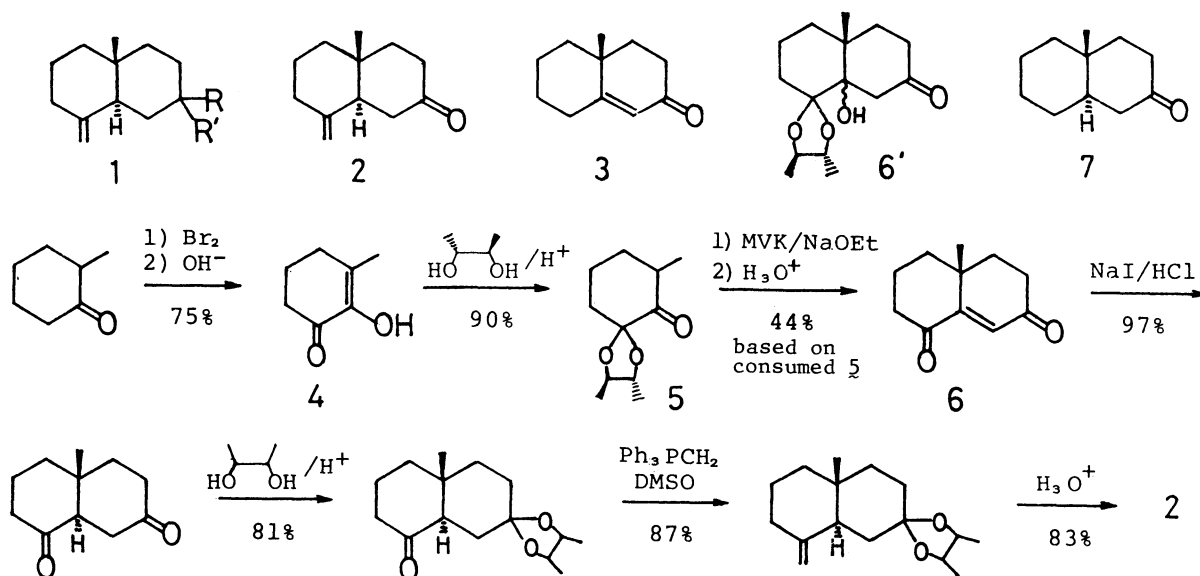
The first total synthesis of eudesmane sesquiterpenes **1** demonstrated by Marshall et al.¹⁾ in 1965 included the titled racemic methylene ketone **2** as the key intermediate which was prepared from 2-methylcyclohexanone and methyl vinyl ketone via an enone **3**. This method has been accepted widely²⁾ and Wijnberg et al. have succeeded in improving the functional group transformation from **3** to **2**.³⁾ An important problem is, however, the fact that the key intermediate **2** has never been obtained in an optically active form. It seems very difficult to conduct an asymmetric Michael addition between the enolate of 2-methylcyclohexanone and methyl vinyl ketone.

We introduce here a novel method for the preparation of **2** by use of 3-methyl-1,2-cyclohexanedione, which is simple and is capable of producing the optically active form. The reaction sequence is shown in Scheme 1.

2-Methylcyclohexanone was converted easily to the enol of 3-methyl-1,2-cyclohexanedione (**4**) in 75% yield according to the method reported.⁴⁾ Acetalization of **4** with (R,R)-2,3-butanediol (1.15 equiv.) was achieved selectively by refluxing in benzene with a minimum amount of *p*-TsOH to give **5** in 90% yield. The α -keto acetal **5** dissolved in NaOEt(0.1 equiv.)/EtOH reacted with methyl vinyl ketone (1.5 equiv.) at -15--5 °C for 23 h to give the annelation product **6'** in 43% yield and the starting **5** in 32% recovery.⁵⁾ Then **6'** was treated with aqueous HCl to give enedione **6** in 69% yield. The asymmetric Michael addition was carried out at various temperatures as shown in Table 1.

The configuration of the carbon-4 α was determined as R after conversion to ketone **7** ($[\alpha]_D -17^\circ$ (c 0.8, CHCl₃)) (lit.⁶⁾ $[\alpha]_D +33^\circ$ (c 1.14, CHCl₃) for the S configuration), which is in agreement with the stereochemistry of asymmetric induction presumed in Fig. 1. The enedione **6** was transformed to **2**⁷⁾ ($[\alpha]_D -23.9^\circ$ (c 1.17, CHCl₃)) in 56% yield according to the sequence shown.

This is the first asymmetric synthesis of optically active **2**, though the % ee remains to be improved. Further elaboration of the asymmetric induction is now in progress in our laboratory.



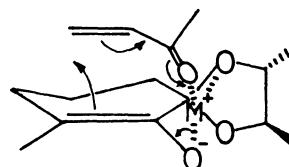
Scheme 1.

Table 1. Asymmetric Michael Addition of α -Keto Acetal 5 with Methyl Vinyl Ketone Leading to Annelation Product 6' and Enedione 6^{a)}

Temp/ $^{\circ}\text{C}$	Time/h	6'		5	6		
		Yield/%	$[\alpha]_{\text{D}}^{\circ}$ (c) ^{b)}		Recovd/%	Yield ^{c)} /%	$[\alpha]_{\text{D}}^{\circ}$ (c) ^{b)}
25	2.5	78	-5.4 (1.00)	12	83	+24.0 (1.15)	30
-5-15	23	64	-5.0 (1.21)	19	69	+34.7 (1.17)	42
-15--5	23	43	-4.2 (1.01)	32	69	+39.0 (0.97)	45
-15--10	48	29	-5.4 (1.03)	39	75	+35.1 (0.81)	42

a) Methyl vinyl ketone 1.5 equiv.; NaOEt (0.1 equiv.) / EtOH. b) In CHCl_3 . c) The yield for 6' \rightarrow 6. d) Determined by ^1H NMR using the signal of =CH- with $\text{Eu}(\text{hfc})_3$.

Fig. 1. The asymmetric induction in Michael addition. The (R) configuration is formed.



References

- 1) J. A. Marshall, M. T. Pike, and R. D. Carroll, *J. Org. Chem.*, **31**, 2933 (1966).
- 2) C. H. Heathcock in "The Total Synthesis of Natural Products," ed by J. Apsimon, Wiley, New York (1973), Vol. 2, Chap. 2.
- 3) J. B. P. A. Wijnberg, J. Vader, and A. de Groot, *J. Org. Chem.*, **48**, 4380 (1983).
J. B. P. A. Wijnberg, G. Jongedijk, and A. de Groot, *Ibid.*, **50**, 2650 (1985).
- 4) M. Utaka, S. Matsushita, and A. Takeda, *Chem. Lett.*, **1980**, 779.
- 5) For the Michael addition of 1,2-cyclohexanediones, see M. Utaka, Y. Fujii, and A. Takeda, *Chem. Lett.*, **1985**, 1123.
- 6) B. Riniker, J. Kalvoda, D. Arigoni, A. Furst, O. Jeger, A. M. Gold, and R. B. Woodward, *J. Am. Chem. Soc.*, **76**, 313 (1954).
- 7) The IR and ^1H NMR spectral data were identical with those reported in Ref. 1.

(Received April 18, 1986)